

Chemistry: WebElements Periodic Table: Professional Edition: Lead: compound data [lead (II) oxide]

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Lead

82

Pb

207.2(1)

Compounds of lead:

lead (II) oxide

- **Formula as commonly written:** PbO
- **Hill system formula:** O₁Pb₁
- **CAS registry number:** [1317-36-8]
- **Formula weight:** 223.199
- **Class:** oxide

Synonyms

- lead (II) oxide
- lead oxide
- litharge

Physical properties

- **Colour:** red or yellow (massicot)
- **Appearance:**
- **Melting point:** 888°C
- **Boiling point:**
- **Density:** 9350 kg m⁻³

Element analysis and oxidation numbers

For each compound, and where possible, a formal oxidation number for each element is given, but the usefulness of this number is limited, especially so for *p*-block elements in particular. Based upon that oxidation number, an electronic configuration is also given but note that for more exotic compounds you should view this as a guide only.

Element	%	Formal oxidation state	Formal electronic configuration



c n
Sele
belo
Flu

Electronegativities

O 7.17 -2

[He].2s².2p⁶ Pl**Effective nuclear charges**

Pb 92.83 2

[Xe].4f¹⁴.5d¹⁰.6s² Pl**Electron binding energies****Synthesis****Chl.****Atom radii**

Not available Pl

Valence shell radii**Solid state structure**

Pl

physical propertiesBulk properties
(density, resistivity,
etc.)

- **Geometry of lead:**
- **Prototypical structure:**

Bro

Pl

Thermal properties
(melting point, etc.)

lead(II) oxide Pl

Thermodynamic
properties**Iod**

Pl

crystallography

Crystal structure

[view VR world]

[view pdb image]

Hyc

Pl

nuclear properties

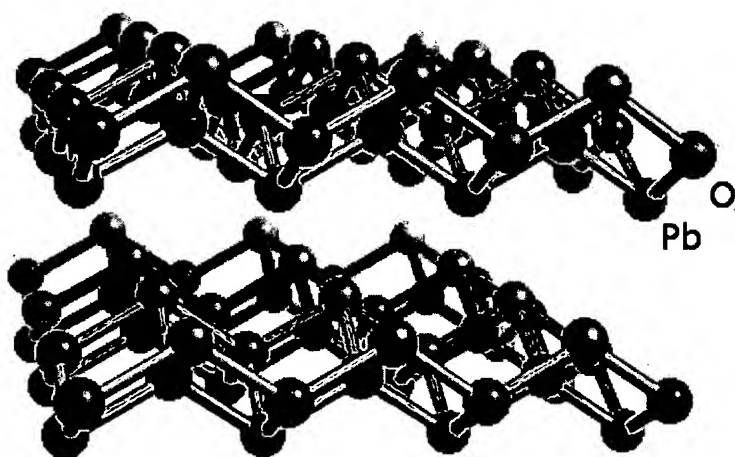
NMR

Oxi.

Pl

Pl

Pl




Naturally occurring isotopes


Radioisotopes

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Isotope pattern

What follows is the calculated isotope pattern for the PbO unit with the most intense ion set to 100%.

Formula: Pb₁O₁

mass	%
220	2.7
221	0.0
222	45.9
223	42.1
224	100.0
225	0.1
226	0.2

PI

Sul

PI

Sel

PI

Tell

PI

Nitr

n

Suppliers

Coming soon....

References

The data on these compounds pages are assembled and adapted from the primary literature and several other sources including the following.


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